

Material Safety Data Sheet

Mercury(II) chloride, reagent ACS, crystals

MSDS# 00943

Section 1 - Chemical Product and Company Identification

MSDS Name: Mercury(II) chloride, reagent ACS, crystals

Catalog Numbers:

AC423930000, AC423930025, AC423930050, AC423935000, 42393-1000

Synonyms:

Calochlor; Corrosive mercury chloride; Corrosive sublimate; Mercury bichloride; Mercury perchloride;

Mercury(II) chloride; Mercuric chloride.

Acros Organics BVBA Company Identification:

Janssen Pharmaceuticalaan 3a

2440 Geel, Belgium

Acros Organics

Company Identification: (USA) One Reagent Lane

Fair Lawn, NJ 07410

For information in the US, call: 800-ACROS-01

For information in Europe, call: +32 14 57 52 11

Emergency Number, Europe: +32 14 57 52 99

Emergency Number US: 201-796-7100

CHEMTREC Phone Number, US: 800-424-9300

CHEMTREC Phone Number, Europe: 703-527-3887

Section 2 - Composition, Information on Ingredients

CAS#: 7487-94-7

Chemical Name: Mercury(II) chloride

%:

100

EINECS#: 231-299-8

Hazard Symbols:

T+CN







28 34 48/24/25 50/53



Section 3 - Hazards Identification

EMERGENCY OVERVIEW

Danger! May be fatal if swallowed. Harmful if inhaled. May cause allergic skin reaction. May be fatal if absorbed through the skin. Light sensitive. May cause kidney damage. May cause central nervous system effects. Causes digestive and respiratory tract irritation with possible burns. Causes severe eye and skin irritation with possible burns. May cause harm to the unborn child. May impair fertility. Severe marine pollutant. Target Organs: Kidneys, central nervous system, reproductive system.

Potential Health Effects

Exposure to mercury or mercury compounds can cause discoloration on the front surface of the lens, which does not interfere with vision. Causes severe eye irritation and possible burns. Contact with mercury or mercury Eye:

compounds can cause ulceration of the conjunctiva and cornea.

May be fatal if absorbed through the skin. Causes severe skin irritation and possible burns. May cause allergic Skin:

contact dermatitis

May be fatal if swallowed. Causes gastrointestinal irritation with nausea, vomiting and diarrhea. Causes Ingestion: gastrointestinal tract burns. May cause muscle tremor and impaired motor function. May cause cardiac

disturbances.

May cause central nervous system effects including vertigo, anxiety, depression, muscle incoordination, and emotional instability. May cause gastrointestinal effects including gum and mouth inflammation, jaw necrosis, and Inhalation:

loosening of the teeth. May cause burns to the respiratory tract. Acute exposure to high concentrations of

mercury vapors may cause severe respiratory tract irritation.

Prolonged or repeated skin contact may cause dermatitis. Chronic inhalation and ingestion may cause effects similar to those of acute inhalation and ingestion. May cause reproductive and fetal effects. Chronic ingestion may Chronic: cause accumulation of mercury in body tissues. Laboratory experiments have resulted in mutagenic effects. May

be rapidly transferred across the placenta and cause adverse fetal effects.

Section 4 - First Aid Measures

In case of contact, immediately flush eyes with plenty of water for at least 15 minutes. Get medical aid Eyes:

immediately.

In case of contact, immediately flush skin with plenty of water for at least 15 minutes while removing Skin:

contaminated clothing and shoes. Get medical aid immediately, Wash clothing before reuse.

POISON material. If swallowed, get medical aid immediately. Only induce vomiting if directed to do so by Ingestion:

medical personnel. Never give anything by mouth to an unconscious person.

If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Inhalation:

Get medical aid.

The concentration of mercury in whole blood is a reasonable measure of the body-burden of mercury and Notes to thus is used for monitoring purposes. Persons with kidney disease, chronic respiratory disease, liver disease, Physician:

or skin disease may be at increased risk from exposure to this substance.

The use of Dimercaprol or BAL (British Anti-Lewisite) as a chelating agent should be determined by qualified

medical personnel. The use of d-Penicillamine as a chelating agent should be determined by qualified medical

personnel.

Antidote:

Section 5 - Fire Fighting Measures

As in any fire, wear a self-contained breathing apparatus in pressure-demand, MSHA/NIOSH (approved General or equivalent), and full protective gear. Water runoff can cause environmental damage. Dike and collect Information:

water used to fight fire. During a fire, irritating and highly toxic gases may be generated by thermal

decomposition or combustion. Substance is noncombustible.

Extinguishing Use water fog, dry chemical, carbon dioxide or alcohol type foam. Media:

Autoignition Not available. Temperature:

Flash Point: Not applicable.

Explosion Not available Limits: Lower:

Explosion Not available Limits: Upper:

Spills/Leaks:

NFPA Rating: health: 4; flammability: 0; instability: 0;

Section 6 - Accidental Release Measures

General Use proper personal protective equipment as indicated in Section 8. Information:

Vacuum or sweep up material and place into a suitable disposal container. Avoid runoff into storm sewers and ditches which lead to waterways. Clean up spills immediately, observing precautions in the Protective

Equipment section. Avoid generating dusty conditions. Provide ventilation.

Section 7 - Handling and Storage

Wash thoroughly after handling. Wash thoroughly after handling. Remove contaminated clothing and wash before reuse. Minimize dust generation and accumulation. Do not breathe dust, mist, or vapor. Do not get in eyes, on

Handling: skin, or on clothing. Do not ingest or inhale. Store protected from light. Use only with adequate ventilation. Extreme care should always be taken to prevent skin and gastrointestinal absorption because these routes of entry can greatly increase the total body burden and are often overlooked in occupational settings.

Store in a tightly closed container. Keep away from food and drinking water. Store in a cool, dry, well-ventilated

Storage:

area away from incompatible substances. Store protected from light.

Section 8 - Exposure Controls, Personal Protection

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1	Chemical Name	ACGIH	NIOSH	OSHA - Final PELs
	Mercury(II) chlorid e	(as Hg) (listed under Mercury inorganic compounds).Skin potential significant	O.05 mg/m3 TWA (vapor, except organo alkyls, as organo alkyls, as organo alkyls, as organo alkyls, as organo orga	none listed

OSHA Vacated PELs: Mercury(II) chloride: None listed

Engineering Controls:

Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower. Use adequate general or local exhaust ventilation to keep airborne concentrations below the permissible exposure limits.

Exposure Limits

Personal Protective Equipment

Eyes: Wear appropriate protective eyeglasses or chemical safety goggles as described by OSHA's eye and face

protection regulations in 29 CFR 1910.133 or European Standard EN166.

Skin: Wear appropriate protective gloves to prevent skin exposure.

Clothing: Wear appropriate protective clothing to prevent skin exposure.

Respirators: A respiratory protection program that meets OSHA's 29 CFR 1910.134 and ANSI Z88.2 requirements or European Standard EN 149 must be followed whenever workplace conditions warrant respirator use.

Section 9 - Physical and Chemical Properties

Physical State: Crystals

Color: white Odor: odorless

pH: 4.7

Vapor Pressure: Not available

Vapor Density: Not available

Evaporation Rate: Negligible.

Viscosity: Not applicable.

Boiling Point: 300 deg C (572.00°F)

Freezing/Melting Point: 277 deg C (530.60°F)

Decomposition Temperature: Not available

Solubility in water: Soluble

Specific Gravity/Density: 5.44 at 25°C

Molecular Formula: HgCl2 Molecular Weight: 271.50

Section 10 - Stability and Reactivity

Chemical Stability: Stable at room temperature in closed containers under normal storage and handling conditions.

Conditions to Avoid: Light, dust generation, excess heat.

Strong oxidizing agents, strong bases, ammonia, copper, iron, silver salts, potassium, antimony, Incompatibilities with sodium, lead, hypophosphites, formates, sulfites, phosphates, albumin, gelatin, alkalies, alkaloid salts,

Other Materials lime water, arsenic, bromides, borax, carbonates, reduced iron, infusions of cinchona, columbo, oak

Section 11 - Toxicological Information

bark or senna, tannic acid, metallic halides, vegetable astringents.

Hazardous

Decomposition

Mercury/mercury oxides, chloride fumes.

Products

Hazardous Polymerization

Will not occur.

RTECS#:

CAS# 7487-94-7: OV9100000

RTECS:

CAS# 7487-94-7: Draize test, rabbit, eye: 50 ug/24H Severe;

Draize test, rabbit, skin: 500 mg/24H Severe;

LD50/LC50:

Oral, mouse: LD50 = 6 mg/kg; Oral, rat: LD50 = 1 mg/kg: Skin, rat: LD50 = 41 mg/kg;

Carcinogenicity:

Mercury(II) chloride - IARC: Group 3 (not classifiable)

Other:

See actual entry in RTECS for complete information.

Section 12 - Ecological Information

Fish: Rainbow trout: LC50 = 0.903 mg/L; 24 Hr; Unspecified

Fish: Fathead Minnow: LC50 = 0.037 mg/L; 48 Hr; Unspecified

Ecotoxicity:

Fish: Bluegill/Sunfish: LC50 = 0.16 mg/L; 96 Hr; Static at 13.5-16.2°C (pH 7.1-7.3)

Water flea Daphnia: LC50 = 0.093 mg/L; 48 Hr; Unspecified

Section 13 - Disposal Considerations

Dispose of in a manner consistent with federal, state, and local regulations.

Section 14 - Transport Information

US DOT

Shipping Name: MERCURIC CHLORIDE

Hazard Class: 6.1 UN Number: UN1624 Packing Group: II Canada TDG

Shipping Name: Not available

Hazard Class: **UN Number:** Packing Group:

Section 15 - Regulatory Information

European/International Regulations

European Labeling in Accordance with EC Directives

Hazard Symbols: T+ C N

Risk Phrases:

R 28 Very toxic if swallowed.

R 34 Causes burns.

R 48/24/25 Toxic: danger of serious damage to health by prolonged exposure in contact with skin and if swallowed.

R 50/53 Very toxic to aquatic organisms, may cause long-term adverse effects in the aquatic environment.

Safety Phrases:

S 36/37/39 Wear suitable protective clothing, gloves and eye/face protection.

S 45 In case of accident or if you feel unwell, seek medical advice immediately (show the label where possible).

S 60 This material and its container must be disposed of as hazardous waste.

S 61 Avoid release to the environment. Refer to special instructions/safety data sheets.

WGK (Water Danger/Protection)

CAS# 7487-94-7: 3

Canada

CAS# 7487-94-7 is listed on Canada's DSL List

Canadian WHMIS Classifications: D1A, D2A, E

This product has been classified in accordance with the hazard criteria of the Controlled Products Regulations and the MSDS contains all of the information required by those regulations.

CAS# 7487-94-7 is listed on Canada's Ingredient Disclosure List

US Federal

TSCA

CAS# 7487-94-7 is listed on the TSCA Inventory.

REVIEWED

Section 16 - Other Information MSDS Creation Date: 6/15/1999

Revision #5 Date 7/20/2009

DATE: July 18/2012

The information above is believed to be accurate and represents the best information currently available to us. However, we make no warranty of merchantibility or any other warranty, express or implied, with respect to such information, and we assume no liability resulting from its use. Users should make their own investigations to determine the suitability of the information for their particular purposes. In no event shall the company be liable for any claims, losses, or damages of any third party or for lost profits or any special, indirect, incidental, consequential, or exemplary damages howsoever arising, even if the company has been advised of the possibility of such damages.